COMPLEX MOLECULES TOWARD LOW-MASS PROTOSTARS: THE SERPENS CORE*

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ABSTRACT

Gas-phase complex organic molecules are commonly detected toward high-mass protostellar hot cores. Detections toward low-mass protostars and outflows are comparatively rare, and a larger sample is the key to investigate how the chemistry responds to its environment. Guided by the prediction that complex organic molecules form in CH₃OH-rich ices and thermally or non-thermally evaporate with CH₃OH, we have identified three sight lines in the Serpens core—SMM1, SMM4, and SMM4-W—which are likely to be rich in complex organics. Using the IRAM 30 m telescope, narrow lines (FWHM of 1–2 km s⁻¹) of CH₃CHO and CH₃OCH₃ are detected toward all sources, HCOOCH₃ toward SMM1 and SMM4-W, and C₂H₅OH not at all. Beam-averaged abundances of individual complex organics range between 0.6% and 10% with respect to CH₃OH when the CH₃OH rotational temperature is applied. The summed complex organic abundances also vary by an order of magnitude, with the richest chemistry toward the most luminous protostar SMM1. The range of abundances compare well with other beam-averaged observations of low-mass sources. Complex organic abundances are of the same order of magnitude toward low-mass protostars and high-mass hot cores, but HCOOCH₃ is relatively more important toward low-mass protostars. This is consistent with a sequential ice photochemistry, dominated by CHO-containing products at low temperatures and early times.

Key words: astrobiology – astrochemistry – ISM: abundances – ISM: molecules

Online-only material: color figures

1. INTRODUCTION

Complex organic molecules in space are traditionally associated with hot cores in high-mass protostars. IRAS 16293–2422 seems to be a low mass equivalent to the high-mass hot cores (van Dishoeck et al. 1995; Cazaux et al. 2003), and there are at least three other low-mass protostars where complex molecules, such as HCOOCH₃ and CH₃OCH₃, are detected (e.g., Bottinelli et al. 2004; Jørgensen et al. 2005). A low-mass outflow has also been detected in a number of organic species (Arce et al. 2008), and HCOOCH₃ and CH₃CHO were recently detected in a cold and quiescent part of a low-mass star-forming region that probably has been illuminated by UV through an outflow cavity (Öberg et al. 2010). Complex organic molecules are clearly present in a range of environments, illustrating the presence of efficient pre-biotic pathways to chemical complexity (see Herbst & van Dishoeck 2009 for a review).

To explain the observed abundances of complex organics (1%-50%) with respect to CH_3OH), and their formation in environments with orders of magnitude differences in density, temperature, radiation flux, and age, most models invoke grain surface chemistry. Garrod & Herbst (2006) and Garrod et al. (2008) set up a model framework, where complex molecules form in ices from recombination of photodissociation fragments of simpler species, especially CH_3OH ice. The fragments become mobile in the protostellar phase as the grains are falling in toward the protostars, forming larger and larger molecules that eventually evaporate. This scheme can reproduce many

aspects of the molecular abundance patterns observed in hot-core regions. An ice formation scenario is also consistent with a hot-core survey (Bisschop et al. 2007), where most complex molecules are closely correlated with the grain surface product CH₃OH.

Laboratory experiments confirm that UV-induced CH₃OH ice chemistry is a feasible way of producing large amounts of the observed hot-core molecules (e.g., Gerakines et al. 1995; Öberg et al. 2009a). The experiments also reveal that non-thermal diffusion of radicals may be possible in the ice before the onset of thermal diffusion, resulting in a cold $T < 20~\rm K$ production channel of complex organics. Since UV photodesorption is efficient (Öberg et al. 2009b, 2009c), many gas-phase complex molecules toward protostars may be present in the envelope or in outflow cavities, and not only in the innermost hot core where ices are thermally evaporated.

Low-mass star-forming regions are less complicated than the high-mass equivalents and are therefore a good test bed for theories on complex molecule formation. Because of the small sample of low-mass sources (six total) with complex molecule detections, it has so far been difficult to ascertain (1) whether there is a real difference between the chemical evolution in low- and high-mass protostars and (2) if the predictions coming out of laboratory experiments and models are supported by observations. The Serpens molecular core is a prime target to search for complex molecules and thus expands the current sample of low-mass complex molecule sources. Several of the protostars in the core are embedded in material where CH₃OH makes up 30% of the ice with respect to H₂O, the most CH₃OHice-rich low-mass region observed to date (Pontoppidan et al. 2004). From CH₃OH gas observations (Kristensen et al. 2010), the region experiences significant ice desorption, which is a requirement to actually observe the products of the complex grain surface chemistry. Similar conditions toward B1-b were

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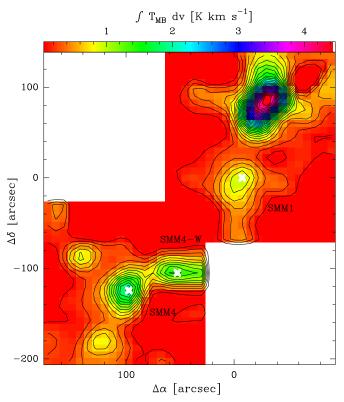


Figure 1. Integrated intensity map of CH₃OH 7₀–6₀ A⁺ line at 338.41 GHz. The SMM4, SMM4-W, and SMM1 positions are marked with white crosses (Kristensen et al. 2010).

(A color version of this figure is available in the online journal.)

previously used to justify a successful search for complex molecules there (Öberg et al. 2010).

Within the Serpens core we targeted the lines of sight toward the protostars SMM1 and SMM4 and toward SMM4-W, an outflow associated with SMM4. SMM1 and SMM4 have bolometric luminosities of $\sim \! \! 30$ and 5 L_{\odot} , respectively, assuming a distance of 250 pc (van Dishoeck et al. 2011). These three lines of sight thus sample three distinctly different environments within the same core, with potentially different chemical evolutionary tracks. This offers a prime opportunity to address the response of the complex organic chemistry to the environment and thus provide insight into their dominating formation mechanisms.

The paper is structured as follows. Section 2 presents the observational parameters for the search of complex organics toward SMM1, SMM4, and SMM4-W. The search results are used together with CH₃OH column densities from Kristensen et al. (2010) to derive abundances with respect to CH₃OH in Section 3. The determined abundances are discussed with respect to the different environments within the Serpens core and the distribution of complex molecules in low- and highmass protostellar regions more generally in Section 4. Based on these comparisons we then discuss the formation mechanisms of complex organic molecules.

2. OBSERVATIONS AND DATA REDUCTION

Figure 1 shows a map of CH₃OH emission adapted from Kristensen et al. (2010) with SMM1, SMM4, and SMM4-W marked. The three sources were observed with the IRAM 30 m telescope in 2010 July, during good weather conditions ($\tau \sim 0.2$ –0.3), using the Eight MIxer Receiver (EMIR). The three

frequency setups in Table 1 were designed to cover a maximum number of complex organic molecules, typically detected in hot cores, with the focus on lines with upper level energies of 10–100 K because of the previously discovered low excitation temperature of CH₃OH toward the same sources. The settings contain multiple lines for HCOOCH₃, CH₃CHO, CH₃OCH₃, and C₂H₅OH (and additional HOCH₂CHO and C₂H₅CN lines) to allow us to constrain the excitation temperatures of any detected species. The line frequencies are taken from the JPL molecular database and the Cologne Database for Molecular Spectroscopy (Müller et al. 2001).

Each receiver setting consists of a combination of the EMIR 90 and 150 GHz receivers (Table 1). At these wavelengths, the beam sizes are \sim 28" and 17", respectively. All three settings were used toward SMM1 and SMM4, while only settings 1 and 2 were observed toward SMM4-W.

Each receiver was connected to a unit of the autocorrelator, with a spectral resolution of 40 kHz and a bandwidth of 120 MHz, equivalent to an unsmoothed velocity resolution of \sim 0.1 km s⁻¹. All spectra were then smoothed to 0.3–0.5 km s⁻¹ to enhance the signal-to-noise ratio. Typical system temperatures were 100–150 K. All observations were carried out using wobbler switching with a 100" throw. Pointing was checked every \sim 2 hr on Mars, J1749+096, or J1741–038 with a typical pointing accuracy of 1".5. All intensities reported in this paper are expressed in units of main-beam brightness temperature, which were converted from antenna temperatures in *pako* using reported main beam and forward efficiencies ($B_{\rm eff}$ and $F_{\rm eff}$) of 75% and 95% with the E90 receiver and 69% and 93% with E150 receiver. The rms in mK are reported in Table 1 for the binned spectra. We assume a calibration uncertainty of 10%.

The data were reduced with the CLASS program, part of the GILDAS software package (see http://www.iram.fr/IRAMFR/GILDAS). Linear (first-order) baselines were determined from velocity ranges without emission features, and then subtracted from the spectra. Some velocity channels showed spikes (Versatile SPectrometer Array (VESPA) spikes are a known problem, in particular in conjunction with EMIR), which were replaced by interpolating the closest two good channels. There were no spikes within 10 km s⁻¹ of any targeted line and replacing the bad channels does not affect the results.

3. RESULTS

This section presents detected complex molecule spectra and the inferred abundances with respect to CH₃OH. First, the CH₃OH column densities and rotational temperatures are extracted from a previously published CH₃OH map convolved to the E90 and E150 beam sizes. Second, complex organic lines are identified from the new observations, and integrated line intensities are derived. Third, the complex molecular column densities and abundances are calculated assuming that the level population of the complex molecules can be described by the excitation temperatures derived for CH₃OH.

3.1. CH₃OH Columns and Temperatures

 CH_3OH observations toward the Serpens core are presented in Kristensen et al. (2010) in the form of a James Clerk Maxwell Telescope (JCMT) HARP-B map. To compare CH_3OH and complex molecule abundances, the fully sampled CH_3OH map, obtained with a 15" beam, in Figure 1 was convolved using MIRIAD (Sault et al. 1995) with the complex molecule beam sizes of \sim 17" and 28" for the 2 and 3 mm observations,

Table 1 Source Characteristics and Observational Parameters

Name	R.A. (J2000)	Decl. (J2000)	$L_{ m bol}/L_{\odot}$
SMM1	18:29:49.80	01:15:20.5	30
SMM4	18:29:56.60	01:13:15.1	5
SMM4-W	18:29:11.60	01:13:18.1	

Complex molecule observations $\delta V \, (\text{km s}^{-1})$ Receiver Frequency Band (GHz) rms (mK) Setting SMM1 SMM4 SMM4-W E90 88.816-88.924 7 7 0.53 10 8 E150 130.836-130.944 11 0.35 2 F90 115.465-115.573 16 18 29 0.41 E150 146.931-147.039 10 0.32 6 6 9 10 0.44 E90 106.631-106.739 . . . 9 E150 146.817-146.923 11 0.32

Table 2 CH₃OH Column Densities and Rotational Temperatures for Different

Source	$N/10^{14}$	cm ⁻²	Rotational T/K			
	17"	28"	17"	28"		
SMM1	2.5 ± 0.3	2.5 ± 0.3	15.9 ± 0.4	15.1 ± 0.4		
SMM4	11.5 ± 1.0	9.4 ± 1.0	12.7 ± 0.2	12.4 ± 0.2		
SMM4-W	22 ± 7	15 ± 4	10.4 ± 0.5	10.8 ± 0.5		

respectively. The resulting CH₃OH line intensities at the SMM1, SMM4, and SMM4-W positions are used to derive new column densities and rotational temperatures using the rotational diagram method (see the Appendix). The column densities and excitation temperatures are listed in Table 2, demonstrating that the applied beam size has only a marginal impact. CH₃OH thus seems equally abundant on 17" and 28" scales, suggesting that the emission is dominated by the protostellar envelope rather than a central hot core.

The derived column densities toward SMM1 and SMM4-W are up to 30% lower compared to the results presented in Kristensen et al. (2010) based on a 15" beam. This is due to underweighting of some weak lines in Kristensen et al. (2010), which resulted in higher rotational temperatures and higher column densities.

3.2. Complex Molecule Line Detections

Figure 2 shows the complete spectra for all settings. Figure 3 zooms in on the most diagnostic lines and shows that SMM1, SMM4, and SMM4-W all contain complex organic molecules with peak intensities of 20-100 mK. CH₃CHO and CH₃OCH₃ are detected in all lines of sight, while at least two HCOOCH₃ lines are detected toward SMM1 and SMM4-W and tentatively detected toward SMM4. C₂H₅OH, HCOCH₂OH, and C₂H₅CN are not detected.

Table 3 lists all detections and significant upper limits. The FWHM vary between 0.7 and 2.5 toward SMM1, 1.0 and 2.6 km s^{-1} toward SMM4, and $1.6 \text{ and } 4.4 \text{ km s}^{-1}$ toward SMM4-W. Only the HCOOCH3 lines around 88.8 GHz and the CH₃CHO line around 115 GHz are not blended, however, and these lines have FWHM of 0.7–1.2, 1.0, and 1.6–2.3 km s⁻¹ toward SMM1, SMM4, and SMM4-W, respectively. These lines are narrow compared to the CH₃OH FWHM of 3.4–4.0 km s⁻¹ toward all three sources (Kristensen et al. 2010). There is thus

no evidence that the complex organics originate in a hotter or more turbulent region than CH₃OH, justifying our presentation of beam-averaged abundances below.

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The line positions of all complex molecules toward SMM1 and SMM4 are consistent with a constant source velocity, indicating that all molecular emission has the same origin. This is not the case toward SMM4-W. The observed HCOOCH₃ emission has the same source velocity as SMM4, while the CH₃CHO and the CH₃OCH₃ emission are shifted by \sim 5 km s⁻¹, comparable to the shift in CH₃OH emission observed by Kristensen et al. (2010). CH₃CHO and CH₃OCH₃ emission then probably originate in the observed CH₃OH outflow, while HCOOCH₃ emission originates in a more quiescent region associated with the extended SMM4 envelope.

The lines are integrated numerically assuming central line velocities of 7, 7, and 7-12 km s⁻¹, for SMM1, SMM4, and SMM4-W, respectively. 3σ upper limits are derived assuming 1 km s⁻¹ line widths toward SMM1 and SMM4 and 2 km s⁻¹ line widths toward SMM4-W, and the standard equation $\sigma =$ rms \times FWHM/ $\sqrt{n_{\rm ch}}$, where $n_{\rm ch}$ is the number of channels across the FWHM. The rms and resolutions are listed in Table 1 and the resulting line intensities and upper limits in Table 3. The lowest upper limits are $\sim 20 \text{ mK km s}^{-1}$, while detections range between 30 and 560 mK km s⁻¹; most upper limits are significant compared to the detected line intensities.

3.3. Complex Molecule Column Densities and Abundances

Assuming that the molecules are thermalized at a rotational temperature, optically thin emission, and no beam dilution, we can use

$$N = \frac{1.67 \times 10^{14}}{\nu \mu^2 S} Q(T_{\text{rot}}) e^{-E_u/T_{\text{rot}}} \int T_b dv,$$
 (1)

from Thi et al. (2004), where N is the total number of molecules, $Q(T_{\rm rot})$ is the temperature-dependent partition function, $E_{\rm u}$ is the energy of the upper level in \bar{K} , T_{rot} is the rotational temperature, and $\int T_b dv$ is the integrated line intensity. $\mu^2 S$ is gathered from Cologne Database for Molecular Spectroscopy (CDMS) for each molecule or calculated from I(300 K) in JPL, $Q(T_{\text{rot}})$ is calculated at the CH₃OH rotational temperature by secondorder interpolation between the three closest values, and $E_{\rm u}$ is calculated from the listed E_{low} values. Using the CH₃OH excitation temperature is a reasonable approximation since the

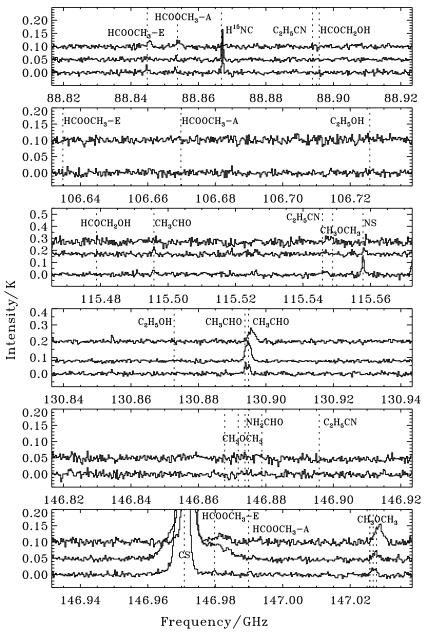


Figure 2. Complete IRAM 30 m spectra toward the SMM4-W outflow (upper spectra), SMM4 (middle spectra), and SMM1 (lower spectra) in the three different E90/E150 settings. The spectra toward SMM4 and SMM-W are offset for clarity. In addition to the labeled lines, there is an unidentified line at 130.855 GHz in SMM-4 W.

thermal desorption temperatures and the non-thermal desorption efficiencies are expected to be similar for CH₃OH and more complex ices. Also, because of the fact that $\mu^2 S$ is similar for all considered molecules, they should be excited to a similar level.

The resulting column densities are listed in Table 4 and when constraints are available from multiple lines, the column densities are consistent within a factor of two. This supports the assumption that the CH_3OH gas and the complex molecules have a similar excitation temperature. The abundances of complex molecules are calculated based on the derived complex molecule column densities and the CH_3OH column densities listed in Table 2. SMM1 has the highest overall complex molecular abundances with 10% HCOOCH₃, 6% CH₃CHO, and 5% CH₃OCH₃, all with respect to CH₃OH. In the other two sources most complex abundances are $\sim 1\%$ with respect to CH₃OH.

4. DISCUSSION

4.1. Variations in the Complex Molecule Distributions in the Serpens Core

The three observed positions in the Serpens core are different both with respect to their total complex molecular content, as traced by $HCOOCH_3$, CH_3CHO , and CH_3OCH_3 , and the relative abundances of the targeted molecules. The summed-detected complex molecular abundance with respect to CH_3OH is $\sim 22\%$ toward SMM1, 3% toward SMM4, and 5% toward SMM4-W. The abundances thus vary over almost an order of magnitude, with the most complex molecules toward the most luminous protostar. This is in line with expectations that, averaged over the envelope, more complex molecules are produced in the CH_3OH ices when a larger portion of the envelope is heated and exposed to more intense UV radiation.

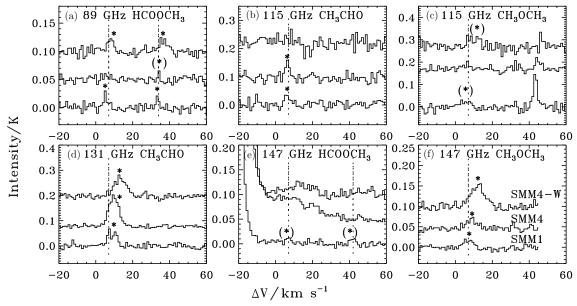


Figure 3. HCOOCH₃, CH₃CHO, and CH₃OCH₃ lines toward the SMM4-W outflow (upper spectra), SMM4 core (middle spectra), and SMM1 core (lower spectra). Detections are marked with stars and tentative detections with stars in parentheses. The dotted line marks the SMM1 velocity of 7 km s⁻¹. The spectra toward SMM4 and SMM-W are offset for clarity.

 ${\bf Table~3}$ Observed Complex Molecular Lines with 1σ Uncertainties in Brackets and 3σ Upper Limits

Molecule	Transition	Freq.	E_{low} (cm ⁻¹)	FWHM/km s ⁻¹			$\int T dV/\text{mK km s}^{-1}$		
		(GHz)		SMM1	SMM4	SMM4-W	SMM1	SMM4	SMM4-W
HCOOCH ₃ -E	7 ₁₆ -7 ₁₅	88.8432	9.5	1.1[0.6]		1.6[0.3]	66[7]	<21	99[8]
HCOOCH ₃ -A	$7_{16} - 7_{15}$	88.8516	9.5	0.7[0.5]		2.3[0.5]	33[7]	<21	116[8]
cis-CH ₂ OHCHO	$9_{46} - 9_{37}$	88.8925	21.2				<21	<21	<24
HCOOCH ₃ -E	$14_{4\ 11} - 14_{3\ 12}$	106.6328	47				<27	< 30	
HCOOCH ₃ -A	$14_{4\ 11}$ $-14_{3\ 12}$	106.6681	47				<27	< 30	
CH ₃ CHO-A	6_{25} -5_{24}	115.4939	15.9	1.2[0.9]	1.0[0.2]		120[20]	120[20]	< 78
C ₂ H ₅ OH	5 ₁₅ -5 ₀₅	115.4945	48.2				< 50	<60	< 78
CH ₃ OCH ₃	$5_{152} - 5_{042}$	115.5440	6.3				<150	<90	<156
CH ₃ OCH ₃	$5_{153} - 5_{043}$	115.5440	6.3						
CH ₃ OCH ₃	$5_{151} - 5_{041}$	115.5448	6.3						
CH ₃ OCH ₃	$5_{150} - 5_{040}$	115.5457	6.3						
C ₂ H ₅ OH	$4_{3}_{2}_{2}$ $-4_{2}_{3}_{2}$	130.8715	9.3				< 30	<24	<27
CH ₃ CHO-E	7_{17} – 6_{16}	130.8918	14.7	2.5[0.3]	2.6[0.1]	3.1[0.2]	290[10]	560[10]	150[9]
CH ₃ CHO-A	7_{17} – 6_{16}	130.8927	14.7						
CH ₃ CH ₂ CN	15_{015} – 14_{014}	130.9039	30.9				< 30	<24	<27
CH ₃ OCH ₃	5332-5242	146.8660	13.4				<27	<33	
CH ₃ OCH ₃	5333-5243	146.8703	13.4						
CH ₃ OCH ₃	5331-5241	146.8725	13.4						
CH ₃ OCH ₃	$5_{330} - 5_{240}$	146.8773	13.4						
HCOOCH3-E	123 10-113 9	146.9777	31.3				<63	<490	<140
HCOOCH ₃ -A	123 10-113 9	146.9880	31.2				<18	<35	< 30
CH ₃ OCH ₃	7 ₁₇₂ -6 ₀₆₂	147.0242	13.2	2.4	2.2	4.4	84[6]	52[6]	120[10]
CH ₃ OCH ₃	7 ₁₇₃ -6 ₀₆₃	147.0242	13.2				0.[0]	22[0]	120[10]
CH ₃ OCH ₃	7 ₁₇₁ -6 ₀₆₁	147.0249	13.2						
CH ₃ OCH ₃	7 ₁₇₀ -6 ₀₆₀	147.0256	13.2						

Toward SMM1 and SMM4-W, HCOOCH₃ is the most abundant complex molecule, while CH₃CHO dominates the observed chemistry toward SMM4. In a scenario where most complex molecules originate from UV chemistry in CH₃OH-rich ices, both HCOOCH₃ and CH₃CHO are expected to mainly form when cold, CO-rich ices are irradiated (Öberg et al. 2009a). In contrast, CH₃OCH₃ is only expected to form efficiently from CH₃OH ice photochemistry once CO has evaporated. The HCOOCH₃/CH₃OCH₃ ratio is therefore a potential tracer

of the relative importance of cold versus lukewarm complex ice chemistry. The (HCOOCH $_3$ +CH $_3$ CHO)/CH $_3$ OCH $_3$ abundance ratios are ~ 3 toward all three sources, providing evidence for the cold complex chemistry dominating the beamaveraged emission—consistent with the derived low rotational temperatures. Gas observations of ice chemistry products in such cold regions are possible due to, e.g., UV photodesorption, desorption due to release of chemical energy, and sputtering in shocks.

 Table 4

 Derived Column Densities and Abundances Using CH3OH Rotational Temperatures

Molecule	Transition	Beam Size		$N/10^{13} \text{ cm}^{-2}$:		% CH ₃ OH	
			SMM1	SMM4	SMM4-W	SMM1	SMM4	SMM4-W
HCOOCH ₃ -E	7 ₁₆ -7 ₁₅	28	1.7	< 0.5	2.4	6.7	< 0.5	1.6
HCOOCH3-A	$7_{16} - 7_{15}$	28	0.8	< 0.5	2.9	3.3	< 0.5	1.9
HCOOCH3-A	$12_{3\ 10}$ $-11_{3\ 9}$	17	<1.4	<4.6	< 7.2	< 5.7	<4.0	< 3.3
CH ₃ CHO-A	6_{25} -5_{24}	21	0.9	1.0	1.0	3.7	0.9	0.4
CH ₃ CHO-E	7 ₁₇ -6 ₁₆	21	0.7	1.5	0.5	2.9	1.3	0.2
CH ₃ CHO-A	7 ₁₇ -6 ₁₆	19	0.7	1.5	0.5	2.9	1.3	0.2
CH ₃ OCH ₃	$5_{152} - 5_{042}$	21	<2.1	<1.2	< 2.0	< 8.4	<1.0	< 0.9
CH ₃ OCH ₃	5 ₁₅₃ -5 ₀₄₃							
CH ₃ OCH ₃	$5_{151} - 5_{041}$							
CH ₃ OCH ₃	$5_{150} - 5_{040}$							
CH ₃ OCH ₃	$5_{332} - 5_{242}$	17	< 0.9	< 1.2		< 3.7	<1.1	
CH ₃ OCH ₃	$5_{333} - 5_{243}$							
CH ₃ OCH ₃	$5_{331} - 5_{241}$							
CH ₃ OCH ₃	$5_{330} - 5_{240}$							
CH ₃ OCH ₃	7_{172} – 6_{062}	17	1.3	0.9	2.4	5.3	0.8	1.1
CH ₃ OCH ₃	$7_{173} - 6_{063}$							
CH ₃ OCH ₃	$7_{171} - 6_{061}$							
CH ₃ OCH ₃	7_{170} – 6_{060}							
C_2H_5OH	$4_{3\ 2\ 2}$ $-4_{2\ 3\ 2}$	19	< 0.9	< 0.6	< 0.8	<3.4	< 0.6	< 0.3

Based on the observations, the main desorption mechanisms may differ for different complex molecules close to outflows. The low central velocities of the HCOOCH₃ lines toward the outflow SMM4-W suggests that it does not originate at the shock front where ices are sputtered by the shock, but rather in the surrounding colder material, where UV rays from the outflow cavity may still be able to penetrate. In contrast, the CH₃CHO and CH₃OCH₃ emission coincides with the outflow (traced by, e.g., CH₃OH). Similarly toward B1-b, HCOOCH₃ has no outflow signature, while CH₃CHO seems partially associated with a weak outflow (Öberg et al. 2010).

This difference between HCOOCH₃, CH₃CHO, and CH₃OCH₃ emission may be the result of different ice surface and ice mantle compositions together with the experimental results that UV photodesorption is only active in the ice surface layer (Öberg et al. 2009b, 2009c), while the shock front results in desorption of the entire ice mantle. If HCOOCH₃ mainly forms on ice surfaces, low levels of HCOOCH₃ gas should be present throughout the protostellar envelopes. Its relative importance to other complex molecules should be less in regions where the entire ice mantle has desorbed at, e.g., shock fronts and inside hot cores. Determining the mechanism behind the different ice surface and ice mantle compositions requires detailed modeling beyond the scope of this paper. Below we instead aim to provide more empirical constraints on the ice chemistry evolution by comparing our main results with previous observations toward low- and high-mass protostars.

4.2. Comparison with Previous Complex Chemistry Observations

Table 5 lists the low-mass sources toward which at least one hot-core-type molecule has been detected. The list comprises one illuminated cold core edge, two outflows, six protostellar envelopes, and three low-mass hot cores observed with 2" beams. Most previous single-dish studies assumed some level of beam dilution based on the assumption that the complex molecules mainly originate from thermal desorption close to the protostar. Öberg et al. (2010) showed that abundant HCOOCH₃ may instead be due to non-thermal desorption at large scales,

and a priori it is difficult to tell whether the complex organics observed with single-dish telescopes toward low-mass young stellar objects mainly reside in a hot core or are spread out in the envelope. We have therefore used abundances derived under the assumption that CH₃OH and complex organics have the same distributions (Herbst & van Dishoeck 2009).

The summed-detected complex molecule abundances are 2.5%–6% toward the outflows (B1-b, SMM4-W, L1157), 9%–56% toward the protostellar envelopes (excluding SMM4), where emission from the cold and hot regions are blended, and 2%-3% toward the low-mass hot cores IRAS 16293A, IRAS 16293B, and IRAS2A for which interferometer data exist. The complex organics abundance is generally higher in protostellar envelopes compared to in outflows. This suggests a chemical evolution where more CH₃OH ice has been converted into complex organics in the protostellar envelope compared to the more pristine ices that are released by the outflow shocks and radiation (assuming the main function of the shocks is to desorb ices rather than initiate chemistry). The SMM4 abundances are more similar to the outflow sources than envelopes around the other, much more luminous protostars, which may be due to the fact that too small an envelope region is affected by the protostar in this case to affect the observable chemistry.

The low-mass hot cores should be further chemically evolved than the envelopes since the ices are expected to evolve during infall from the colder parts of the envelope. It is therefore unexpected that the complex abundances with respect to CH₃OH are smaller in the angularly resolved observations. As discussed in Section 4.1, different ice surface and ice mantle compositions may play a large role in the observed gas-phase abundance ratios. Assuming that complex molecules form mainly in the ice surface layer in cold regions, their abundances with respect to CH₃OH in the surface layer should increase with radiation dose, e.g., between the outflows and the protostar. As long as non-thermal desorption regulates the gas-phase abundances (which should be true in most of the protostellar envelope), it is only the complexorganics-enriched ice surface layer that is reflected in the gas-phase abundances. Once thermal desorption becomes the dominant mechanism to release CH₃OH and complex organics

 Table 5

 CH₃OH Data and Beam-averaged Complex Molecule Abundances Toward Low-mass Sources

Source		CH ₃ OH		$HCOOCH_3$	CH_3CHO	CH ₃ OCH ₃	C_2H_5OH	Refs.		
	$N/10^{14} \text{ cm}^{-2}$	Beam/"	$T_{\rm rot}/{ m K}$	% СН ₃ ОН						
B1-b core	4.7	19	10[5]	2.3	1.2	< 0.8	<1.0	1		
SMM4-W	22[7]	22.5	11[1]	3.5	0.6	1.1	< 0.4			
L1157 outflow	15	10.5	12[2]	1.8			0.7	2, 3		
SMM1	2.5[0.3]	22.5	16[1]	10.0	6.6	5.3	< 3.4			
SMM4	10.5[1.0]	22.5	13[1]	<1.0	2.2	0.8	< 0.6			
NGC1333 IRAS 4Aa	5.1	10	24	56		<22		4, 5, 6		
NGC1333 IRAS 4B ^a	3.5	10	34	26		<19		4, 5, 6		
IRAS 16293 env.a	8.8	20	85	30	4	20		6, 7, 8		
IRAS 16293 Ab	1.1×10^{4}	$1''.3 \times 2''.7$	100	0.6	< 0.02	0.6	1.4	9, 10		
IRAS 16293 Bb	5.0×10^{3}	$1''.3 \times 2''.7$	100	0.8	0.6	1.6		9, 10		
NGC1333 IRAS 2Ab	2×10^4	1".8 × 1".0	>75 K			2		11, 12		

Notes.

References. (1) Öberg et al. 2010; (2) Bachiller & Perez Gutierrez 1997; (3) Arce et al. 2008; (4) Maret et al. 2005; (5) Bottinelli et al. 2007; (6) Herbst & van Dishoeck 2009; (7) van Dishoeck et al. 1995; (8) Cazaux et al. 2003; (9) Kuan et al. 2004; (10) Bisschop et al. 2008; (11) Huang et al. 2005; (12) Jørgensen et al. 2005.

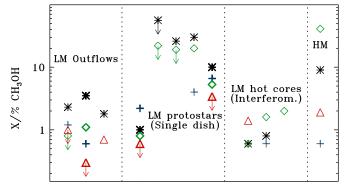


Figure 4. Comparison of HCOOCH $_3$ (black stars), CH $_3$ CHO (blue crosses), CH $_3$ OCH $_3$ (green diamonds), and C $_2$ H $_5$ OH (red triangles) abundances with respect to CH $_3$ OH toward low-mass outflows (B1-b, SMM4-W, L1157), protostars observed with a single dish (SMM4, IRAS 4A, IRAS 4B, IRAS 16293, SMM1) and protostars observed interferometrically (IRAS 16293A, IRAS 16293B, IRAS 2A), and the average high-mass hot-core abundances from Bisschop et al. (2007).

(A color version of this figure is available in the online journal.)

into the gas phase, the gas-phase abundances instead reflect the complex organics to $\mathrm{CH_3OH}$ ratio in the bulk of the ice mantle, which may be much lower. To confirm this scenario clearly requires a combination of laboratory experiments on surface versus bulk chemistry, astrochemical modeling, and more hotcore interferometric observations.

In the unresolved observations that probe both cold and warm material, $HCOOCH_3$ is generally the most abundant complex molecule (except for CH_3CHO toward SMM 4) while CH_3OCH_3 or C_2H_5OH is the most abundant molecule in the low-mass hot-core observations. This supports an ice chemistry scenario where complex molecules form sequentially starting with HCO-rich molecules ($HCOOCH_3$ and/or CH_3CHO) as long as CO ice is abundant, followed by $CH_{3/2}$ -rich molecules at higher ice temperatures.

Figure 4 also shows how the low-mass sample of complex molecule abundances compare with average values from a single-dish survey toward high-mass hot cores (Bisschop et al. 2007). Generally, the abundances are quite similar within a

factor of a few toward low- and high-mass sources, but the low-mass sources do generally have lower CH₃OCH₃ abundances compared to the hot-core sample. While the difference in absolute numbers may be due to the lack of applied beam dilution toward the low-mass sources, the average HCOOCH₃/CH₃OCH₃ ratio toward the high-mass hot cores of 0.2 is an order of magnitude lower than the ratio toward the low-mass sources observed with a single-dish telescope. The molecules in the high-mass hot core have a higher temperature chemical history. The difference between the low-mass and high-mass sample further supports that the formation of complex organics is a sequential process, starting with HCO-rich organics at low temperatures.

5. CONCLUSIONS

- 1. Two low-mass protostars (SMM1 and SMM4) and a low-mass outflow (SMM4-W) in the Serpens core, known for its high CH₃OH ice and gas abundances, have been probed for complex organic molecules. HCOOCH₃, CH₃CHO, and CH₃OCH₃ are detected at beam-averaged abundances of 0.4%–10% with respect to CH₃OH.
- 2. The complex molecule abundances with respect to CH₃OH vary by an order of magnitude, with the richest complex chemistry toward the most luminous source. (HCOOCH₃+CH₃CHO)/CH₃OCH₃ abundance ratios are ~3, suggesting that cold ice chemistry dominates the complex molecule production.
- 3. The beam-averaged abundances and excitation temperature limits indicate that the complex organics are distributed over large scales with low densities and/or low kinetic temperatures. This implies that most complex molecular emission in these sources originate from non-thermal desorption of cold ices rather than a hot core.
- 4. Comparison with previous observations reveals that HCOOCH₃ generally dominates the complex chemistry toward outflows and low-mass protostellar envelopes, consistent with experiments on CO:CH₃OH ice photochemistry below 25 K. In contrast, observations of high-mass and resolved low-mass hot cores contain more CH₃OCH₃ or C₂H₅OH than HCOOCH₃. This supports a sequential

^a The complex abundances values were calculated assuming the *same* beam dilution for CH₃OH and the complex species, which should correspond to comparing beam-averaged abundances since the beams are of similar size (Herbst & van Dishoeck 2009).

^b The complex abundances are also from interferometric observations.

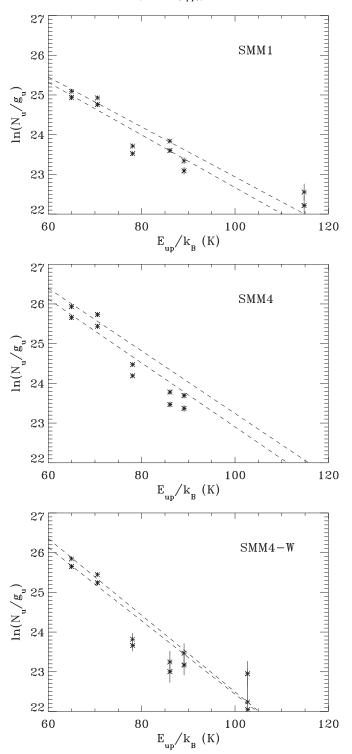


Figure 5. CH_3OH rotational diagrams toward SMM1, SMM4, and SMM4-W for a 17'' beam (upper points) and a 28'' beam (lower points).

formation of complex molecules, starting with HCO-rich molecules as long as CO ice is abundant, followed by $\text{CH}_{3/2}$ -rich molecules at higher ice temperatures.

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APPENDIX

CH₃OH ROTATIONAL DIAGRAMS

Figure 5 shows the rotational diagrams used to derive CH₃OH rotational temperatures and beam-averaged column densities for 17" and 28" beams (Goldsmith & Langer 1999). The integrated CH₃OH intensities are from the maps presented in Kristensen et al. (2010) and all other values from the JPL spectral database and CDMS. The integrated main-beam temperatures are related to the column density in the upper energy level by

$$\frac{N_{\rm u}}{g_{\rm u}} = \frac{3k \int T_{\rm MB} dV}{8\pi^3 \nu \mu^2 S},\tag{A1}$$

where $N_{\rm u}$ is the column density of the upper level, $g_{\rm u}$ is the degeneracy in the upper level, k is Boltzmann's constant, ν is the transition frequency, μ is the dipole moment, and S is the line strength. The total beam-averaged column density $N_{\rm T}$ in cm⁻² can then be computed from

$$\frac{N_{\rm u}}{g_{\rm u}} = \frac{N_{\rm T}}{Q(T_{\rm rot})} e^{-E_{\rm u}/T_{\rm rot}},\tag{A2}$$

where $Q(T_{\rm rot})$ is the rotational partition function and $E_{\rm u}$ is the upper level energy in K. In the rotational diagram, the logarithm of $N_{\rm u}/g_{\rm u}$ is plotted versus $E_{\rm u}$ and the column density is derived from the y-axis intercept and the temperature from the slope. Figure 5 shows that the CH₃OH lines are reasonably well fit by a single temperature in all lines of sight when taking into account the line intensity uncertainties and the uncertainties in relative CH₃OH A and E abundances.

There are, however, some systematic deviations from these fits, implying that the level populations cannot be characterized by single excitation temperatures perfectly. This is expected since material at different distances from the protostars and from the shock fronts are included in the beam. Kristensen et al. (2010) modeled the CH₃OH abundances toward several Serpens cores using both the rotational diagram method and using RATRAN and envelope models with radially symmetric temperature and density profiles. The obtained CH₃OH abundances were remarkably similar, within a factor of two. For the purpose of this study, where the CH₃OH abundances are only needed as a reference for the abundances of more complex molecules, the rotational diagram method should therefore be sufficient.

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